Highly Diastereoselective Thioglycosylation of Functionalized Peracetylated Glycosides Catalyzed by $MoO₂Cl₂$

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Received September 27, 2006

ABSTRACT

Among 18 oxometallic species, MoO2Cl2 was found to be the most reactive in catalytic thioglycosylation of O-acetylated glycosides with functionalized thiols in CH2Cl2, leading cleanly to 1,2-trans-thioglycosides with exclusive diastereocontrol. The new catalytic protocol is applicable to a monoglycoside building block and *â***-(1**f**6)-S-linked-thiodisaccharide synthesis.**

(Oligo)thioglycosides serve as key mediators for the recognition of cellular surface and receptor enzymes.¹ In addition, they are important O -glycan mimetics to study guest-host pairing events of glycoproteins in protein transport, cell-tocell interaction, and signal transduction.2 Surfactant-type thioglycosides have also been applied to tertiary structural re-constitution of deactivated membrane proteins bearing cysteine residues.3 More importantly, monosaccharides possessing glycosidic sulfhydryl (SH) and arylthio (SAr) groups are versatile donors in carbohydrate synthesis and function as universal building blocks and key precursors in oligosaccharide synthesis.4 By taking advantage of the differential reactivity of various thioglycosides, many desired sequences of oligosaccharides have been delicately constructed in one pot5 in the presence of suitable glycosidic bond-forming promoters.6

Conventional methods of appending thioglycoside units to peracetylated sugars with thiols require the use of

stoichiometric, moisture-sensitive catalysts such as ZnX_2 , $SnCl₄$,⁸ FeCl₃,⁹ TiCl₄-InCl₃,¹⁰ BF₃-Et₂O,¹¹ and TMSOTf^{12a} with moderate to good 1,2-*trans* selectivity. Notably, the catalytic application of TMSOTf mandates the use of airand moisture-sensitive $Me₃SiSMe^{12b}$ or $Bu₃SnSEt^{12c}$ as the only thiol source. In addition, a one-pot iodination/thiogly-

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cosylation of peracetylated sugars catalyzed by I_2 /HMDS (60 mol % each) in the presence of MeSSMe or thiols has been demonstrated.¹³ So far, only $ZrCl_4$ can mediate thioglycosylation of peracetylated glycosides with complete 1,2-*trans* selectivity at 0° C in stoichiometric sense.¹⁴ However, only a 1:1 mixture of α/β anomers was obtained when the reaction was carried out at ambient temperature. Therefore, a new mild, water-tolerant, and stereoselective catalyst to fulfill such a transformation remains to be explored.

As part of our ongoing projects by using vanadyl and oxometallic species in catalyzing $C-C$ and $C-X$ bond formation,¹⁵ asymmetric aerobic oxidative dehydrogenation¹⁶ and coupling,¹⁷ and photoinitiated DNA cleavage,¹⁸ we evaluated the feasibility of catalyzing the thioglycosylation event by using amphoteric and water-tolerant oxometallic species. Herein we disclose our preliminary finding toward this end.

We started out by using penta- O -acetyl- β -D-glucose 1 and thiophenol (1.3 equiv) as a test thioglycosylation system with a diverse array of oxometallic species $(1-5 \text{ mol } \%)$ in optimal solvent $CH₂Cl₂$ at ambient temperature. Among 18 different oxometallic species examined,¹⁹ MoO₂Cl₂ (1 mol %) was found to be the most reactive and efficient catalyst, leading only to the β -anomer 2a in 88% yield along with the recovery of the starting α -anomer 1 (7-8%). Some of the β -anomer 1 is isomerized into α -anomer 1 during the catalytic conditions, suggesting the involvement of an oxocarbenium ion intermediate. Notably, no desired product was obtained when $MoCl₅$ was employed as the catalyst, indicating the participating role of the oxometallic unit(s) in $MoO₂Cl₂$. Molybdenum species have been recognized as competent oxidative sulfur-transfer catalysts in episulfide formation of strained cyclic alkenes and transsulfidation of isonitriles ($RN\equiv C$) to isothiocyanates $(RN\equiv C\equiv S)$.^{20a,b} Notably, nonoxidative thiol group transfer of the current study has never been documented.^{20c,d}

With the optimal thioglycosylation catalyst $MoO₂Cl₂$ in hand, we started to look at the substrate scope with aromatic and aliphatic thiols of varying steric and electronic demands. For the given β -anomer 1, the current catalytic protocol is

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- (19) See the Supporting Information for details.

applicable to three different 4- and 2-substituted-benzenethiols (Table 1). In general, the sterically encumbered 2-sub-

^a Isolated yields after chromatographic purification. *^b* 5 mol % of catalyst was used. *^c* 3 equiv of *tert*-butylmercaptan was used.

stituted-benzenethiols are less reactive (20 h) and lower yielding $(47-64\%$ yields) than the corresponding 4-substituted ones (16 h; 86-90% yields). In addition, thiols bearing electron-donating substituents (e.g., $CH₃$ and $OCH₃$) are less reactive (16 h) and lower yielding (86-90% yields) than those (10 h and 94-95% yields) bearing electron-withdrawing substituents (e.g., Cl and benzo-fused 2-Np). Notably, about 44% and 32% of the isomerized substrate (i.e., α -anomer 1) was recovered respectively in the cases of slower reacting 2-methyl- and 2-methoxybenzenethiols which are responsible for the lower isolated yields in the products **2f** and **2g**.

Four straight alkanethiols of varying steric attributes were also examined. The rates of catalytic thioglycosylation were significantly decreased up to a factor of 8 with increasing steric demands of the thiols (i.e., Et > PhCH₂ > c-C₆H₁₁ > *t*-Bu). In addition, the chemical yields dropped from 88% to 71%. Functionalized alkanethiols like 11-mercaptoundecanoate and 2-*tert*-butyldimethylsiloxyethanethiol (TBSO- $(CH₂)₂SH$) are also suitable substrates for the β -thioglycosylation, leading to **2m** and **2n** in satisfactory yields (64- 71%). Notably, monothioglycosylation can be achieved for dithiols like 1,3-propanedithiol. The expected **2***l* can be isolated in 68% yield, thus allowing for subsequent functional group manipulation at the remaining thiol unit in **2***l*. Since all the recovered starting α -anomer 1 may be readily converted back to the original β -anomer 1 and the catalyst can be recovered from the aqueous layer, the current catalytic protocol is unprecedented and meets the standard of green chemistry.

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To gain insights into the origin of 1,2-*trans* diastereocontrol, 2-deoxy-2-phthalimido-*â*-D-glucose was also examined. It was found that exclusive *â*-thioglycosylation by benzenethiol under the optimal catalytic conditions was also observed, Table 2. This result suggests that anchimeric

^a Isolated yields after chromatographic purification. *^b* Performed at 0 °C.

assistance by the neighboring C2-acyl groups is essential for the exclusive 1,2-*trans* selectivity in the thioglycosylation.

Several peracetylated *â*-D-monosaccharides derived from galactose, mannose, ribose, xylose, and arabinose were further investigated. In all cases except the mannose derivative, the thioglycosylations by benzenethiol were complete in $12-14$ h leading to the corresponding β -1-phenylthiosugars **⁴** and **⁶**-**⁸** in 86-92% yields. In the mannose case, the corresponding α -1-phenylthiomannose **5** was obtained in 74% yield within 20 h. Remarkably, the thioglycosylation of 3,4,6-tri-*O*-acetyl-D-glycal by PhSH was complete in 15 min at 0 °C and led to Ferrier-type product **9** with high α -selectivity (α/β > 19/1) presumably due to the preferred axial attack to avoid $A^{1,2}$ -strain.²¹

Peracetylated dissaccharides derived from maltose and lactose were also suitable substrates for the highly diastereoselective thioglycosylation protocol. Exclusive formation of the corresponding β -thioglycosides **10** and **11** (94–95%) yields) with intact $1\rightarrow4'$ glycosidic stereochemistry were achieved in 14 h.

Thioglycosides protected with 4,6-benzylidene acetals are important precursors to highly functionalized saccharides.²² Their conventional synthesis requires the initial preparation of 2,3,4,6-*O*-tetraacetylated monosaccharides followed by complete deacetylation under basic conditions (e.g., NaOMe/ MeOH). We can now access a representative target **12** from D-glucose by using $MoO₂Cl₂$ as the sole catalyst for a fourstep, peracetylation-thioglycosylation-deacetylation acetal formation sequence in $70-75%$ overall yield without purification of any intermediates, Scheme 1. Notably, the 4,6-

benzylidene acetal **12** constitutes a precursor leading to a universal, fully functionalized, glycoside building block **13** for oligosaccharide synthesis.22

As a final demonstration for its practical application in β -(1–6)-*S*-linked-thiodisaccharide synthesis, we treated $D-1,2,3,4,6$ -penta-*O*-acetyl- β -glucose 1 with methyl 2,3,4tri- O -benzyl-6-thio- α -glucose 14 under the optimal reaction conditions. The desired β -(1–6)-*S*-linked-thiodisaccharide **15** was furnished in 74% isolated yield with complete retention at both glycosidic centers, Scheme 2.

In conclusion, we have achieved the first successful catalytic use of neutral $MoO₂Cl₂$ for thioglycosylation of *O*-acetylated saccharides with exclusive 1,2-*trans*-diastereocontrols. The integration of various catalytic pathways mediated by $MoO₂Cl₂$ and $VO(OTf)₂$ allows for access of a functional glycoside building block **13**. A representative β -(1–6)-*S*-linked-thiodisaccharide **15** can also be provided

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smoothly by the newly developed catalytic thioglycosylation process, which augurs well for its extensive use in delicate carbohydrate syntheses.

Acknowledgment. We thank the National Science Council of Taiwan for generous financial support of this research.

Supporting Information Available: Spectral data and characterization for products **2a**-**n**, **³**-**12**, **¹⁴**, and **¹⁵**. This material is available free of charge via the Internet at http://pubs.acs.org.

OL062375G